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(54) Title: FUEL ADDITIVE COMPOSITIONS CONTA	INING	ALIPHATIC AMINES AND POLYALKYL HYDROXYAROMATICS
(57) Abstract		
substituted succinimide; and (b) a polyalityl hydroxyaromati	oxyalky ic com	aliphatic amine selected from the group consisting of (1) a straight or l substituted amine, and (3) a straight or branched chain hydrocarbylound or salt thereof wherein the polyalkyl group has sufficient molecular natic compound soluble in hydrocarbons boiling in the gasoline or diesal
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01	FUEL ADDITIVE COMPOSITIONS CONTAINING
02	ALIPHATIC AMINES
03	AND POLYALKYL HYDROXYAROMATICS
04	
05	BACKGROUND OF THE INVENTION
06	
07	This invention relates to a fuel additive composition. More
80	particularly, this invention relates to a fuel additive
09	composition containing an aliphatic amine and a polyalkyl
10	hydroxyaromatic compound.
11	
12	It is well known in the art that liquid hydrocarbon
13	combustion fuels, such as fuel oils and gasolines, tend to
14	exhibit certain deleterious characteristics, either after
15	long periods of storage or under actual operational
16	conditions. Gasolines, for example, in operational use tend
17	to deposit sludge and varnish at various points in the power
18	system, including the carburetor or injectors and the intake
19	valves. It is desirable, therefore, to find a means for
20	improving liquid hydrocarbon fuels by lessening their
21	tendency to leave such deposits.
22	
23	U.S. Patent No. 3,849,085 discloses a motor fuel composition
24	comprising a mixture of hydrocarbon in the gasoline boiling
25	range containing about 0.01 to 0.25 volume percent of a high
26	molecular weight aliphatic hydrocarbon substituted phenol in
27	which the aliphatic hydrocarbon radical has an average
28	molecular weight in the range of about 500 to 3,500. This
29	patent teaches that gasoline compositions containing a minor
30	amount of an aliphatic hydrocarbon substituted phenol not
31	only prevents or inhibits the formation of intake valve and
32	port deposits in a gasoline engine but also enhances the
33	performance of the fuel composition in engines designed to
34	operate at higher operating temperatures with a minimum of

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01 decomposition and deposit formation in the manifold of the 02 engine. 03 04 U.S. Patent No. 4,134,846 discloses a fuel additive 05 composition comprising a mixture of (1) the reaction product 06 of an aliphatic hydrocarbon-substituted phenol, 07 epichlorohydrin and a primary or secondary mono- or polyamine, and (2) a polyalkylene phenol. This patent 80 teaches that such compositions show excellent carburetor, 09 10 induction system and combustion chamber detergency and, in 11 addition, provide effective rust inhibition when used in 12 hydrocarbon fuels at low concentrations. 13 14 U.S. Patent No. 4,231,759 discloses a fuel additive 15 composition comprising the Mannich condensation product of 16 (1) a high molecular weight sulfur-free alkyl-substituted 17 hydroxyaromatic compound wherein the alkyl group has a 18 number average molecular weight of about 600 to 3,000 (2) an 19 amine containing at least one active hydrogen atom, and (3) 20 an aldehyde, wherein the respective molar ratio of reactants 21 is 1:0.1-10 : 0.1-10. 22 23 SUMMARY OF THE INVENTION 24 25 The present invention provides a novel fuel additive 26 composition comprising: 27 28 a fuel-soluble aliphatic amine selected from the group 29 consisting of: 30 31 (1) a straight or branched chain hydrocarbyl-32 substituted amine having at least one basic 33 nitrogen atom wherein the hydrocarbyl group has a 34

-3-

		•
01		number average molecular weight of about 250 to
02		3,000,
03		
04	(2)	a hydroxyalkyl-substituted amine comprising the
05		reaction product of (i) a polyolefin epoxide
06		derived from a branched-chain polyolefin having a
07		number average molecular weight of about 250 to
80		3,000, and (ii) a nitrogen-containing compound
09		selected from ammonia, a monoamine having from 1
10		to 40 carbon atoms, and a polyamine having from 2
11		to about 12 amine nitrogen atoms and from 2 to
12		about 40 carbon atoms, and
13		
14	(3)	a straight or branched chain hydrocarbyl-
15		substituted succinimide comprising the reaction
16		product of a straight or branched chain
17		hydrocarbyl-substituted succinic acid or
18		anhydride, wherein the hydrocarbyl group has a
19		number average molecular weight of about 250 to
20		3,000, and a polyamine having from 2 to about 12
21		amine nitrogen atoms and 2 to about 40 carbon
22		atoms; and .
23		
24		olyalkyl hydroxyaromatic compound or salt thereof
25		rein the polyalkyl group has sufficient molecular
26		ght and carbon chain length to render the polyalkyl
27	-	roxyaromatic compound soluble in hydrocarbons
28	boi	ling in the gasoline or diesel range.
29	•	
30	_	ent invention further provides a fuel composition
31	_	ng a major amount of hydrocarbons boiling in the
32		e or diesel range and an effective detergent amount
33	of the r	novel fuel additive composition described above.

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	·
)1	The present invention is also concerned with a fuel
)2	concentrate comprising an inert stable oleophilic organic
)3	solvent boiling in the range of from about 150°F to 400°F
)4	and from about 10 to 70 weight percent of the fuel additive
)5	composition of the instant invention.
06	
7	Among other factors, the present invention is based on the
8	surprising discovery that the unique combination of an
9	aliphatic amine and a polyalkyl hydroxyaromatic compound
LO	provides unexpectedly superior deposit control performance
11	when compared to each component individually.
L2	Joseph Co data domponent Inday 2000217.
L3	DETAILED DESCRIPTION OF THE INVENTION
L4	ALL ALL VILLE A AVIL VA AND ALL VILLE AVIL
15	The Aliphatic Amine
16	
L7	As noted above, the fuel-soluble aliphatic amine component
18	of the present fuel additive composition is an amine
19	selected from the group consisting of a straight or branched
20	chain hydrocarbyl-substituted amine, a hydroxyalkyl-
21	substituted amine and a hydrocarbyl-substituted succinimide
22	Preferably, such aliphatic amines will be of sufficient
23	molecular weight so as to be nonvolatile at normal engine
24	intake valve operating temperatures, which are generally in
25	the range of about 175°C to 300°.
26	
27	A. The Hydrocarbyl-Substituted Amine
28	
29	The hydrocarbyl-substituted amine employed as the aliphatic
30	amine component of the present fuel additive composition is
31	a straight or branched chain hydrocarbyl-substituted amine
32	having at least one basic nitrogen atom wherein the
33	hydrocarbyl group has a number average molecular weight of
14	about 250 to 3,000.

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Preferably, the hydrocarbyl group will have a number average molecular weight in the range of about 700 to 2,200, and more preferably, in the range of about 900 to 1,500. The hydrocarbyl group may be either straight chain or branched chain. When the hydrocarbyl group is straight chain, a preferred aliphatic amine is oleyl amine.

07

80 When employing a branched chain hydrocarbyl amine, the 09 hydrocarbyl group is preferably derived from polymers of C2 10 to C6 olefins. Such branched-chain hydrocarbyl group will 11 ordinarily be prepared by polymerizing olefins of from 2 to 12 6 carbon atoms (ethylene being copolymerized with another 13 olefin so as to provide a branched-chain). The branched 14 chain hydrocarbyl group will generally have at least 1 15 branch per 6 carbon atoms along the chain, preferably at 16 least 1 branch per 4 carbon atoms along the chain and, more 17 preferably, at least 1 branch per 2 carbon atoms along the 18 chain. The preferred branched-chain hydrocarbyl groups are 19 polypropylene and polyisobutylene. The branches will 20 usually be of from 1 to 2 carbon atoms, preferably 1 carbon 21 atom, that is, methyl. In general, the branched-chain 22 hydrocarbyl group will contain from about 18 to about 214 23 carbon atoms, preferably from about 50 to about 157 carbon 24 atoms.

25 26

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In most instances, the branched-chain hydrocarbyl amines are not a pure single product, but rather a mixture of compounds having an average molecular weight. Usually, the range of molecular weights will be relatively narrow and peaked near the indicated molecular weight.

30 31 32

The amine component of the branched-chain hydrocarbyl amines may be derived from ammonia, a monoamine or a polyamine.

-6-

01 The monoamine or polyamine component embodies a broad class 02 of amines having from 1 to about 12 amine nitrogen atoms and 03 from 1 to 40 carbon atoms with a carbon to nitrogen ratio 04 . between about 1:1 and 10:1. Generally, the monoamine will 05 contain from 1 to about 40 carbon atoms and the polyamine 06 will contain from 2 to about 12 amine nitrogen atoms and 07 from 2 to about 40 carbon atoms. In most instances, the 80 amine component is not a pure single product, but rather a 09 mixture of compounds having a major quantity of the 10 designated amine. For the more complicated polyamines, the 11 compositions will be a mixture of amines having as the major 12 product the compound indicated and having minor amounts of 13 analogous compounds. Suitable monoamines and polyamines are 14 described more fully below in the discussion of 15 hydroxyalkyl-substituted amines. 16 17 When the amine component is a polyamine, it will preferably 18 be a polyalkylene polyamine, including alkylenediamine. 19 Preferably, the alkylene group will contain from 2 to 6 20 carbon atoms, more preferably from 2 to 3 carbon atoms. 21 Examples of such polyamines include ethylene diamine, 22 diethylene triamine, triethylene tetramine and tetraethylene 23 pentamine. Preferred polyamines are ethylene diamine and 24 diethylene triamine. 25 26 A particularly preferred branched-chain hydrocarbyl amine is 27 polyisobutenyl ethylene diamine. 28 29 The branched-chain hydrocarbyl amines employed in the fuel 30 additive composition of the invention are prepared by 31 conventional procedures known in the art. Such branched-32 chain hydrocarbyl amines and their preparations are 33 described in detail in U.S. Patent Nos. 3,438,757; 34

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01 3,565,804; 3,574,576; 3,848,056 and 3,960,515, the
02 disclosures of which are incorporated herein by reference.
03

03

B. The Hydroxyalkyl-Substituted Amine

05

06 The hydroxyalkyl-substituted amine additive employed in the 07 fuel composition of the present invention comprises the 80 reaction product of (a) a polyolefin epoxide derived from a 09 branched chain polyolefin having an average molecular weight 10 of about 250 to 3,000 and (b) a nitrogen-containing compound 11 selected from ammonia, a monoamine having from 1 to 40 12 carbon atoms, and a polyamine having from 2 to about 12 13 amine nitrogen atoms and from 2 to about 40 carbon atoms. 14 The amine component of this reaction product is selected to 15 provide solubility in the fuel composition and deposit 16 control activity.

17 18

Polyolefin Epoxide Component

19 20

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23

The polyolefin epoxide component of the presently employed hydroxyalkyl-substituted amine reaction product is obtained by oxidizing a polyolefin with an oxidizing agent to give an alkylene oxide, or epoxide, in which the oxirane ring is derived from oxidation of the double bond in the polyolefin.

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The polyolefin starting material used in the preparation of the polyolefin epoxide is a high molecular weight branched chain polyolefin having an average molecular weight of about 250 to 3,000, preferably from about 700 to 2,200, and more preferably from about 900 to 1,500.

30 31

Such high molecular weight polyolefins are generally
 mixtures of molecules having different molecular weights and
 can have at least one branch per 6 carbon atoms along the

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chain, preferably at least one branch per 4 carbon atoms 01 along the chain, and particularly preferred that there be 02 about one branch per 2 carbon atoms along the chain. 03 branched chain olefins may conveniently comprise polyolefins 04 prepared by the polymerization of olefins of from 2 to 6 05 carbon atoms, and preferably from olefins of from 3 to 4 06 07 carbon atoms, and more preferably from propylene or isobutylene. When ethylene is employed, it will normally be 80 copolymerized with another olefin so as to provide a 09 branched chain polyolefin. The addition-polymerizable 10 olefins employed are normally 1-olefins. The branch may be 11 12 of from 1 to 4 carbon atoms, more usually of from 1 to 2 13 carbon atoms, and preferably methyl. 14 15 In general, any high molecular weight branched chain 16 polyolefin isomer whose epoxide is capable of reacting with 17 an amine is suitable for use in preparing the presently 18 employed fuel additives. However, sterically hindered 19 epoxides, such as tetra-alkyl substituted epoxides, are 20 generally slower to react. 21 22 Particularly preferred polyolefins are those containing an 23 . alkylvinylidene isomer present in an amount at least about 24 20%, and preferably at least 50%, of the total polyolefin 25 composition. The preferred alkylvinylidene isomers include 26 methylvinylidene and ethylvinylidene, more preferably the 27 methylvinylidene isomer. 28 29 The especially preferred high molecular weight polyolefins 30 used to prepare the instant polyolefin epoxides are 31 polyisobutenes which comprise at least about 20% of the more 32 reactive methylvinylidene isomer, preferably at least 50% and more preferably at least 70%. Suitable polyisobutenes 33 34 include those prepared using BF, catalysts. The preparation WO 94/14929

01 of such polyisobutenes in which the methylvinylidene isomer 02 comprises a high percentage of the total composition is 03 described in U.S. Patent Nos. 4,152,499 and 4,605,808. 04 05 Examples of suitable polyisobutenes having a high alkylvinylidene content include Ultravis 30, a polyisobutene 06 07 having a molecular weight of about 1300 and a 80 methylvinylidene content of about 76%, available from 09 British Petroleum. 10 11 As noted above, the polyolefin is oxidized with a suitable 12 oxidizing agent to provide an alkylene oxide, or polyolefin 13 epoxide, in which the oxirane ring is formed from oxidation 14 of the polyolefin double bond. 15 16 The oxidizing agent employed may be any of the well known 17 conventional oxidizing agents used to oxidize double bonds. 18 Suitable oxidizing agents include hydrogen peroxide, 19 peracetic acid, perbenzoic acid, performic acid, 20 monoperphthalic acid, percamphoric acid, persuccinic acid 21 and petrifluoroacetic acid. The preferred oxidizing agent 22 is peracetic acid. 23 24 When peracetic acid is used as the oxidizing agent, 25 generally a 40% peracetic acid solution and about a 5% 26 equivalent of sodium acetate (as compared to the peracetic 27 acid) is added to the polyolefin in a molar ratio of per-28 acid to olefin in the range of about 1.5:1 to 1:1, 29 preferably about 1.2:1. The mixture is gradually allowed to 30 react at a temperature in the range of about 20°C to 90°C. 31 32 The resulting polyolefin epoxide, which is isolated by 33 conventional techniques, is generally a liquid or semi-solid 34

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01 resin at room temperature, depending on the type and 02 molecular weight of olefin employed. 03 04 Amine Component 05 06 The amine component of the presently employed hydroxyalkyl-07 substituted amine reaction product is derived from a 80 nitrogen-containing compound selected from ammonia, a monoamine having from 1 to 40 carbon atoms, and a polyamine 09 having from 2 to about 12 amine nitrogen atoms and from 2 to 10 11 about 40 carbon atoms. The amine component is reacted with 12 a polyolefin epoxide to produce the hydroxyalkyl-substituted 13 amine fuel additive finding use within the scope of the 14 present invention. The amine component provides a reaction 15 product with, on the average, at least about one basic 16 nitrogen atom per product molecule, i.e., a nitrogen atom 17 titratable by a strong acid. 18 19 Preferably, the amine component is derived from a polyamine 20 having from 2 to about 12 amine nitrogen atoms and from 2 to 21 about 40 carbon atoms. The polyamine preferably has a 22 carbon-to-nitrogen ratio of from about 1:1 to 10:1. 23 24 The polyamine may be substituted with substituents selected 25 from (A) hydrogen, (B) hydrocarbyl groups of from 1 to about 26 10 carbon atoms, (C) acyl groups of from 2 to about 10 27 carbon atoms, and (D) monoketo, monohydroxy, mononitro, 28 monocyano, lower alkyl and lower alkoxy derivatives of (B) 29 and (C). "Lower", as used in terms like lower alkyl or 30 lower alkoxy, means a group containing from 1 to about 6

carbon atoms. At least one of the substituents on one of

a primary or secondary amino nitrogen.

the basic nitrogen atoms of the polyamine is hydrogen, e.g.,

at least one of the basic nitrogen atoms of the polyamine is

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Hydrocarbyl, as used in describing the amine components of 01 this invention, denotes an organic radical composed of 02 carbon and hydrogen which may be aliphatic, alicyclic, 03 aromatic or combinations thereof, e.g., aralkyl. 04 Preferably, the hydrocarbyl group will be relatively free of 05 aliphatic unsaturation, i.e., ethylenic and acetylenic, 06 particularly acetylenic unsaturation. The substituted 07 polyamines of the present invention are generally, but not 80 necessarily, N-substituted polyamines. Exemplary 09 hydrocarbyl groups and substituted hydrocarbyl groups 10 include alkyls such as methyl, ethyl, propyl, butyl, 11 isobutyl, pentyl, hexyl, octyl, etc., alkenyls such as 12 propenyl, isobutenyl, hexenyl, octenyl, etc., hydroxyalkyls, 13 such as 2-hydroxyethyl, 3-hydroxypropyl, hydroxy-isopropyl, 14 4-hydroxybutyl, etc., ketoalkyls, such as 2-ketopropyl, 15 6-ketooctyl, etc., alkoxy and lower alkenoxy alkyls, such as 16 ethoxyethyl, ethoxypropyl, propoxyethyl, propoxypropyl, 17 diethyleneoxymethyl, triethyleneoxyethyl, 18 tetraethyleneoxyethyl, diethyleneoxyhexyl, etc. 19 aforementioned acyl groups (C) are such as propionyl, 20 acetyl, etc. The more preferred substituents are hydrogen, 21 C_1 - C_6 alkyls and C_1 - C_6 hydroxyalkyls. 22 23 In a substituted polyamine, the substituents are found at 24 any atom capable of receiving them. The substituted atoms, 25 26 e.g., substituted nitrogen atoms, are generally geometrically unequivalent, and consequently the substituted 27 amines finding use in the present invention can be mixtures 28 of mono- and poly-substituted polyamines with substituent 29 30 groups situated at equivalent and/or unequivalent atoms. 31 The more preferred polyamine finding use within the scope of 32 the present invention is a polyalkylene polyamine, including 33 34 alkylene diamine, and including substituted polyamines,

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01 e.g., alkyl and hydroxyalkyl-substituted polyalkylene polyamine. Preferably, the alkylene group contains from 2 02 to 6 carbon atoms, there being preferably from 2 to 3 carbon 03 atoms between the nitrogen atoms. Such groups are 04 05 exemplified by ethylene, 1,2-propylene, 2,2-dimethyl-06 propylene, trimethylene, 1,3,2-hydroxypropylene, etc. 07 Examples of such polyamines include ethylene diamine, 80 diethylene triamine, di(trimethylene) triamine, dipropylene triamine, triethylene tetraamine, tripropylene tetraamine, 09 10 tetraethylene pentamine, and pentaethylene hexamine. Such 11 amines encompass isomers such as branched-chain polyamines 12 and previously-mentioned substituted polyamines, including 13 hydroxy- and hydrocarbyl-substituted polyamines. Among the 14 polyalkylene polyamines, those containing 2-12 amino 15 nitrogen atoms and 2-24 carbon atoms are especially 16 preferred, and the C2-C3 alkylene polyamines are most 17 preferred, that is, ethylene diamine, polyethylene 18 polyamine, propylene diamine and polypropylene polyamine, 19 and in particular, the lower polyalkylene polyamines, e.g., 20 ethylene diamine, dipropylene triamine, etc. A particularly 21 preferred polyalkylene polyamine is diethylene triamine. 22 23 The amine component of the presently employed fuel additive 24 also may be derived from heterocyclic polyamines, 25 heterocyclic substituted amines and substituted heterocyclic 26 compounds, wherein the heterocycle comprises one or more 5-6 27 membered rings containing oxygen and/or nitrogen. Such 28 heterocyclic rings may be saturated or unsaturated and 29 substituted with groups selected from the aforementioned 30 (A), (B), (C) and (D). The heterocyclic compounds are 31 exemplified by piperazines, such as 2-methylpiperazine, N-32 (2-hydroxyethyl) -piperazine, 1,2-bis-(N-piperazinyl)ethane 33 and N,N'-bis(N-piperazinyl)piperazine, 2-methylimidazoline, 34 3-aminopiperidine, 3-aminopyridine, N-(3-aminopropyl)-

-13-

01 morpholine, etc. Among the heterocyclic compounds the 02 piperazines are preferred. 03 04 Typical polyamines that can be used to form the additives 05 employed in this invention by reaction with a polyolefin epoxide include the following: ethylene diamine, 06 07 1,2-propylene diamine, 1,3-propylene diamine, diethylene triamine, triethylene tetramine, hexamethylene diamine, 80 tetraethylene pentamine, dimethylaminopropylene diamine, 09 N-(beta-aminoethyl)piperazine, N-(beta-10 11 aminoethyl) piperadine, 3-amino-N-ethylpiperidine, N-(beta-12 aminoethyl) morpholine, N,N'-di(beta-aminoethyl)piperazine, N, N' -di (beta-aminoethyl) imidazolidone-2, N- (beta-cyanoethyl) 13 14 ethane-1,2-diamine, 1-amino-3,6,9-triazaoctadecane, 15 1-amino-3,6-diaza-9-oxadecane, N-(beta-aminoethyl) 16 diethanolamine, N'acetylmethyl-N-(beta-aminoethyl) ethane-1,2-diamine, N-acetonyl-1,2-propanediamine, N-(beta-17 nitroethyl)-1,3-propane diamine, 1,3-dimethyl-5(beta-18 19 aminoethyl) hexahydrotriazine, N-(beta-aminoethyl) -20 hexahydrotriazine, 5-(beta-aminoethyl)-1,3,5-dioxazine, 2-21 (2-aminoethylamino) ethanol, and 2-{2-(2-aminoethylamino) 22 ethylamino] ethanol. 23 24 Alternatively, the amine component of the presently employed 25 hydroxyalkyl-substituted amine may be derived from an amine 26 having the formula: 27 28 29 30 31 wherein R_1 and R_2 are independently selected from the group 32

consisting of hydrogen and hydrocarbyl of 1 to about 20

carbon atoms and, when taken together, R_1 and R_2 may form

33

-14-

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01
     one or more 5- or 6-membered rings containing up to about 20
02
     carbon atoms. Preferably, R_1 is hydrogen and R_2 is a
03
     hydrocarbyl group having 1 to about 10 carbon atoms. More
     preferably, R_1 and R_2 are hydrogen. The hydrocarbyl groups
04
05
     may be straight-chain or branched and may be aliphatic,
06
     alicyclic, aromatic or combinations thereof. The
07
     hydrocarbyl groups may also contain one or more oxygen
80
     atoms.
09
10
     An amine of the above formula is defined as a "secondary
11
     amine" when both R<sub>1</sub> and R<sub>2</sub> are hydrocarbyl. When R<sub>1</sub> is
12
     hydrogen and R2 is hydrocarbyl, the amine is defined as a
13
     "primary amine"; and when both R, and R, are hydrogen, the
14
     amine is ammonia.
15
16
     Primary amines useful in preparing the fuel additives of the
17
     present invention contain 1 nitrogen atom and 1 to about 20
18
     carbon atoms, preferably 1 to 10 carbon atoms. The primary
19
     amine may also contain one or more oxygen atoms.
20
21
     Preferably, the hydrocarbyl group of the primary amine is
22
     methyl, ethyl, propyl, butyl, pentyl, hexyl, octyl, 2-
23
     hydroxyethyl or 2-methoxyethyl. More preferably, the
24
     hydrocarbyl group is methyl, ethyl or propyl.
25
26
     Typical primary amines are exemplified by N-methylamine, N-
27
     ethylamine, N-n-propylamine, N-isopropylamine, N-n-
28
     butylamine, N-isobutylamine, N-sec-butylamine, N-tert-
29
     butylamine, N-n-pentylamine, N-cyclopentylamine, N-n-
30
     hexylamine, N-cyclohexylamine, N-octylamine, N-decylamine,
31
     N-dodecylamine, N-octadecylamine, N-benzylamine, N-(2-
32
     phenylethyl)amine, 2-aminoethanol, 3-amino-1-proponal, 2-(2-
33
      aminoethoxy) ethanol, N-(2-methoxyethyl) amine, N-(2-
34
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-15-

01 ethoxyethyl) amine and the like. Preferred primary amines 02 are N-methylamine, N-ethylamine and N-n-propylamine. 03 04 The amine component of the presently employed fuel additive 05 may also be derived from a secondary amine. The hydrocarbyl 06 groups of the secondary amine may be the same or different 07 and will generally contain 1 to about 20 carbon atoms, 80 preferably 1 to about 10 carbon atoms. One or both of the 09 hydrocarbyl groups may also contain one or more oxygen 10 atoms. 11 12 Preferably, the hydrocarbyl groups of the secondary amine 13 are independently selected from the group consisting of 14 methyl, ethyl, propyl, butyl, pentyl, hexyl, 2-hydroxyethyl 15 and 2-methoxyethyl. More preferably, the hydrocarbyl groups 16 are methyl, ethyl or propyl. 17 Typical secondary amines which may be used in this invention 18 19 include N,N-dimethylamine, N,N-diethylamine, N,N-di-n-20 propylamine, N,N-diisopropylamine, N,N-di-n-butylamine, N,N-21 di-sec-butylamine, N,N-di-n-pentylamine, N,N-di-n-22 hexylamine, N, N-dicyclohexylamine, N, N-dioctylamine, N-23 ethyl-N-methylamine, N-methyl-N-n-propylamine, N-n-butyl-N-24 methylamine, N-methyl-N-octylamine, N-ethyl-N-25 isopropylamine, N-ethyl-N-octylamine, N,N-di(2-26 hydroxyethyl)amine, N,N-di(3-hydroxypropyl)amine, N,N-27 di(ethoxyethyl)amine, N,N-di(propoxyethyl)amine and the like. Preferred secondary amines are N,N-dimethylamine, 28 29 N, N-diethylamine and N, N-di-n-propylamine. 30 31 Cyclic secondary amines may also be employed to form the 32 additives of this invention. In such cyclic compounds, R_1 33 and R_2 of the formula hereinabove, when taken together, form 34

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one or more 5- or 6-membered rings containing up to about 20 .
02
     carbon atoms. The ring containing the amine nitrogen atom
     is generally saturated, but may be fused to one or more
03
     saturated or unsaturated rings.
04
                                      The rings may be
     substituted with hydrocarbyl groups of from 1 to about 10
05
06
     carbon atoms and may contain one or more oxygen atoms.
07
80
     Suitable cyclic secondary amines include piperidine, 4-
09
     methylpiperidine, pyrrolidine, morpholine, 2,6-
10
     dimethylmorpholine and the like.
11
12
     In many instances the amine component is not a single
13
     compound but a mixture in which one or several compounds
14
     predominate with the average composition indicated.
15
     example, tetraethylene pentamine prepared by the
16
     polymerization of aziridine or the reaction of
17
     dichloroethylene and ammonia will have both lower and higher
18
     amine members, e.g., triethylene tetraamine, substituted
19
     piperazines and pentaethylene hexamine, but the composition
20
     will be mainly tetraethylene pentamine and the empirical
21
     formula of the total amine composition will closely
22
     approximate that of tetraethylene pentamine. Finally, in
23
     preparing the compounds of this invention using a polyamine,
24
     where the various nitrogen atoms of the polyamine are not
25
     geometrically equivalent, several substitutional isomers are
26
     possible and are encompassed within the final product.
27
    Methods of preparation of amines and their reactions are
28
     detailed in Sidgewick's "The Organic Chemistry of Nitrogen",
29
     Clarendon Press, Oxford, 1966; Noller's "Chemistry of
30
    Organic Compounds", Saunders, Philadelphia, 2nd Ed., 1957;
31
     and Kirk-Othmer's "Encyclopedia of Chemical Technology", 2nd
32
    Ed., especially Volume 2, pp. 99-116.
33
34
                          Preparation of the
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33 34

used.

Hydroxyalkyl-Substituted Amine Reaction Product 01 02 As noted above, the fuel additive finding use in the present 03 invention is a hydroxyalkyl-substituted amine which is the 04 reaction product of (a) a polyolefin epoxide derived from a 05 branched chain polyolefin having an average molecular weight 06 of about 250 to 3,000 and (b) a nitrogen-containing compound 07 selected from ammonia, a monoamine having from 1 to 40 80 carbon atoms, and a polyamine having from 2 to about 12 09 amine nitrogen atoms and from 2 to about 40 carbon atoms. 10 11 12 The reaction of the polyolefin epoxide and the amine component is generally carried out either neat or with a 13 solvent at a temperature in the range of about 100°C to 14 250°C and preferably from about 180°C to about 220°C. A 15 reaction pressure will generally be maintained in the range 16 from about 1 to 250 atmospheres. The reaction pressure will 17 vary depending on the reaction temperature, presence or 18 absence of solvent and the boiling point of the amine 19 component. The reaction usually is conducted in the absence 20 of oxygen, and may be carried out in the presence or absence 21 of a catalyst. The desired product may be obtained by water 22 23 wash and stripping, usually by aid of vacuum, of any 24 residual solvent. 25 26 The mole ratio of basic amine nitrogen to polyolefin epoxide 27 will generally be in the range of about 3 to 50 moles of basic amine nitrogen per mole of epoxide, and more usually 28 about 5 to 20 moles of basic amine nitrogen per mole of 29 30 epoxide. The mole ratio will depend upon the particular amine and the desired ratio of epoxide to amine. Since 31 32 suppression of polysubstitution of the amine is usually desired, large mole excesses of the amine will generally be

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01 The reaction of polyolefin epoxide and amine may be 02 conducted either in the presence or absence of a catalyst. When employed, suitable catalysts include Lewis acids, such 03 04 as aluminum trichloride, boron trifluoride, titanium 05 tetrachloride, ferric chloride, and the like. Other useful 06 catalysts include solid catalysts containing both Bronsted 07 and Lewis acid sites, such as alumina, silica, silica-80 alumina, and the like. 09 10 The reaction may also be carried out with or without the 11 presence of a reaction solvent. A reaction solvent is 12 generally employed whenever necessary to reduce the 13 viscosity of the reaction product. These solvents should be 14 stable and inert to the reactants and reaction product. 15 Preferred solvents include aliphatic or aromatic 16 hydrocarbons or aliphatic alcohols. 17 18 Depending on the temperature of the reaction, the particular 19 polyolefin epoxide used, the mole ratios and the particular 20 amine, as well as the presence or absence of a catalyst, the 21 reaction time may vary from less than 1 hour to about 72 22 hours. 23 24 After the reaction has been carried out for a sufficient 25 length of time, the reaction mixture may be subjected to 26 extraction with a hydrocarbon-water or hydrocarbon-alcohol-27 water medium to free the product from any low-molecular 28 weight amine salts which have formed and any unreacted 29 polyamines. The product may then be isolated by evaporation 30 of the solvent. 31 32 In most instances, the additive compositions used in this 33 invention are not a pure single product, but rather a 34 mixture of compounds having an average molecular weight.

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Usually, the range of molecular weights will be relatively 01 narrow and peaked near the indicated molecular weight. 02 Similarly, for the more complicated amines, such as 03 polyamines, the compositions will be a mixture of amines 04 having as the major product the compound indicated as the 05 06 average composition and having minor amounts of analogous 07 compounds relatively close in compositions to the dominant 80 compound.

09

C. The Hydrocarbyl-Substituted Succinimide

10 11

12 The hydrocarbyl-substituted succinimide which can be employed as the aliphatic amine component of the present 13 fuel additive composition is a straight or branched chain 14 15 hydrocarbyl-substituted succinimide comprising the reaction 16 product of a straight or branched chain hydrocarbyl-17 substituted succinic acid or anhydride, wherein the hydrocarbyl group has a number average molecular weight of 18 about 250 to 3,000, and a polyamine having from 2 to about 19 20 12 amine nitrogen atoms and 2 to about 40 carbon atoms.

21

Preferably, the hydrocarbyl group will have a number average molecular weight in the range of about 700 to 2,200, and more preferably, in the range of about 900 to 1,500. The hydrocarbyl group may be either straight chain or branched chain. Preferably, the hydrocarbyl group will be a branched chain hydrocarbyl group.

28

When employing a branched chain hydrocarbyl-substituted succinimide, the branched chain hydrocarbyl group is preferably derived from polymers of C₂ to C₆ olefins. Such branched chain hydrocarbyl groups are described more fully above in the discussion of hydrocarbyl-substituted amines and hydroxyalkyl-substituted amines. Preferably, the

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branched chain hydrocarbyl group will be derived from 01 02 polypropylene or polyisobutylene. More preferably, the branched chain hydrocarbyl group will be derived from 03 04 polyisobutylene. 05 06 The succinimides employed in the present invention are 07 prepared by reacting a straight or branched chain 80 hydrocarbyl-substituted succinic acid or anhydride with a 09 polyamine having from 2 to about 12 amine nitrogen atoms and 10 2 to about 40 carbon atoms. 11 Hydrocarbyl-substituted succinic anhydrides are well known 12 13 in the art and are prepared by the thermal reaction of olefins and maleic anhydride as described, for example, in 14 15 U.S. Patent Nos. 3,361,673 and 3,676,089. Alternatively, 16 hydrocarbyl-substituted succinic anhydrides can be prepared by reaction of chlorinated olefins with maleic anhydride as 17 18 described, for example, in U.S. Patent No. 3,172,892. 19 olefin employed in these reactions has a number average 20 molecular weight in the range of about 250 to about 3,000. Preferably, the number average molecular weight of the 21 22 olefin is about 700 to about 2,200, more preferably about 23 900 to 1,500. .24 25 The reaction of a polyamine with an alkenyl or alkyl succinic acid or anhydride to produce a polyamino alkenyl or 26 alkyl succinimide is well known is the art and is described, 27 28 for example, in U.S. Patent Nos. 3,018,291; 3,024,237; 29 3,172,892; 3,219,666; 3,223,495; 3,272,746; 3,361,673 and 30 3,443,918. 31 32

The Amine Component of the Succinimide

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01 The amine moiety of the hydrocarbyl-substituted succinimide 02 is preferably derived from a polyamine having from 2 to 03 about 12 amine nitrogen atoms and from 2 to about 40 carbon 04 atoms. The polyamine is preferably reacted with a 05 hydrocarbyl-substituted succinic acid or anhydride to 06 produce the hydrocarbyl-substituted succinimide fuel 07 additive finding use within the scope of the present 80 invention. The polyamine, encompassing diamines, provides 09 the product succinimide with, on the average, at least about 10 one basic nitrogen atom per succinimide molecule, i.e., a 11 nitrogen atom titratable by strong acid. The polyamine 12 preferably has a carbon-to-nitrogen ratio of from about 1:1 13 to about 10:1. The polyamine may be substituted with 14 substituents selected from hydrogen, hydrocarbyl groups of 15 from 1 to about 10 carbon atoms, acyl groups of from 2 to 16 about 10 carbon atoms, and monoketone, monohydroxy, 17 mononitro, monocyano, alkyl and alkoxy derivatives of 18 hydrocarbyl groups of from 1 to 10 carbon atoms. It is 19 preferred that at least one of the basic nitrogen atoms of 20 the polyamine is a primary or secondary amino nitrogen. 21 polyamine component employed in the present invention has 22 been described and exemplified more fully in U.S. Patent No. 23 4,191,537. 24 25 Hydrocarbyl, as used in describing the amine components used 26 in this invention, denotes an organic radical composed of 27 carbon and hydrogen which may be aliphatic, alicyclic, 28 aromatic or combinations thereof, e.g., aralkyl. 29 Preferably, the hydrocarbyl group will be relatively free of 30 aliphatic unsaturation, i.e., ethylenic and acetylenic, 31 particularly acetylenic unsaturation. The more preferred 32 polyamine finding use within the scope of the present 33 invention is a polyalkylene polyamine, including 34 alkylenediamine, and including substituted polyamines, e.g.,

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alkyl and hydroxyalkyl-substituted polyalkylene polyamine. Preferably, the alkylene group contains from 2 to 6 carbon atoms, there being preferably from 2 to 3 carbon atoms between the nitrogen atoms. Examples of such polyamines include ethylenediamine, diethylene triamine, triethylene tetramine, di(trimethylene) triamine, dipropylene triamine, tetraethylene pentamine, etc. Among the polyalkylene polyamines, polyethylene polyamine and polypropylene polyamine containing 2-12 amine nitrogen atoms and 2-24 carbon atoms are especially preferred and in particular, the lower polyalkylene polyamines, e.g., ethylenediamine, diethylene triamine, propylene diamine, dipropylene triamine, etc., are most preferred. Particularly preferred polyamines are ethylene diamine and diethylene triamine.

The Polyalkyl Hydroxyaromatic Compound

As noted above, the polyalkyl hydroxyaromatic component of the present fuel additive composition is a polyalkyl hydroxyaromatic compound or salt thereof wherein the polyalkyl group has sufficient molecular weight and carbon chain length to render the polyalkyl hydroxyaromatic compound soluble in hydrocarbons boiling in the gasoline or diesel range. As with the aliphatic amine component of the present invention, the polyalkyl hydroxyaromatic compound will preferably be of sufficient molecular weight so as to be nonvolatile at normal engine intake valve operating temperatures, generally in the range of about 175°C to 300°C.

In general, the polyalkyl substituent on the polyalkyl hydroxyaromatic compound will have an average molecular weight in the range of about 400 to 5,000, preferably about 400 to 3,000, more preferably from about 600 to 2,000.

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The polyalkyl-substituted hydroxyaromatic compounds finding 01 use in this invention are derived from hydroxyaromatic 02 hydrocarbons. Such hydroxyaromatic compounds include 03 mononuclear monohydroxy and polyhydroxy aromatic 04 hydrocarbons having 1 to 4, and preferably 1 to 3, hydroxy 05 groups. Suitable hydroxyaromatic compounds include phenol, 06 07 catechol, resorcinol, hydroquinone, pyrogallol, and the like. The preferred hydroxyaromatic compound is phenol. 80 09 Suitable polyalkyl hydroxyaromatic compounds and their 10 preparation are described, for example, in U.S. Patent 11 Nos. 3,849,085; 4,231,759 and 4,238,628, the disclosures of 12 each of which are incorporated herein by reference. 13 14 15 The polyalkyl substituent on the polyalkyl hydroxyaromatic compounds employed in the invention may be generally derived 16 17 from polyolefins which are polymers or copolymers of 18 mono-olefins, particularly 1-mono-olefins, such as ethylene, propylene, butylene, and the like. Preferably, the 19 20 mono-olefin employed will have 2 to about 24 carbon atoms, 21 and more preferably, about 3 to 12 carbon atoms. More 22 preferred mono-olefins include propylene, butylene, particularly isobutylene, 1-octene and 1-decene. 23 24 Polyolefins prepared from such mono-olefins include polypropylene, polybutene, especially polyisobutene, and the 25 polyalphaolefins produced from 1-octene and 1-decene. 26 27 The preferred polyisobutenes used to prepare the presently 28 29 employed polyalkyl hydroxyaromatic compounds are 30 polyisobutenes which comprise at least about 20% of the more 31 reactive methylvinylidene isomer, preferably at least 50% 32 and more preferably at least 70%. Suitable polyisobutenes include those prepared using BF3 catalysts. The preparation 33 34 of such polyisobutenes in which the methylvinylidene isomer

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01 comprises a high percentage of the total composition is 02 described in U.S. Patent Nos. 4,152,499 and 4,605,808. 03 04 Examples of suitable polyisobutenes having a high 05 alkylvinylidene content include Ultravis 30, a polyisobutene 06 having a molecular weight of about 1300 and a 07 methylvinylidene content of about 74%, available from 80 British Petroleum. 09 10 Numerous methods are known for preparing the polyalkyl 11 hydroxyaromatic compounds used in the present invention and 12 any of these are considered suitable for producing the 13 polyalkyl hydroxyaromatic component of the instant fuel additive composition. One such method involves the reaction 14 15 of a phenol with an olefin polymer in the presence of an 16 aluminum chloride-sulfuric acid catalyst, as described in 17 U.S. Patent No. 3,849,085. Similarly, U.S. Patent No. 4,231,759 discloses that polyalkyl hydroxyaromatic 18 19 compounds may be obtained by the alkylation of phenol with 20 polypropylene, polybutylene and other polyalkylene 21 compounds, in the presence of an alkylation catalyst, such 22 as boron trifluoride. 23 24 One preferred method of preparing polyalkyl hydroxyaromatic 25 compounds is disclosed in U.S. Patent No. 4,238,628. 26 patent teaches a process for producing undegraded alkylated 27 phenols by alkylating, at about 0°C to 60°C, a complex 28 comprising boron trifluoride and phenol with a propylene or 29 higher olefin polymer having terminal ethylene units. 30 wherein the molar ratio of complex to olefin polymer is 31 about 1:1 to 3:1. Preferred olefin polymers include 32 polybutene having terminal ethylene units. 33

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01	Preferred polyalkyl hydroxyaromatic compounds finding use in
02	the fuel additive composition of the present invention
03	include polypropylene phenol, polyisobutylene phenol, and
04	polyalkyl phenols derived from polyalphaolefins,
05	particularly 1-decene oligomers.
06	
07	Polyalkyl phenols, wherein the polyalkyl group is derived
80	from polyalphaolefins, such as 1-octene and 1-decene
09	oligomers, are described in PCT International Patent
10	Application Publication No. WO 90/07564, published July 12,
11	1990, the disclosure of which is incorporated herein by
12	reference. This publication teaches that such polyalkyl
13	phenols may be prepared by reacting the appropriate
14	polyalphaolefin with phenol in the presence of an alkylating
15	catalyst at a temperature of from about 60°C to 200°C,
16	either neat or in an inert solvent at atmospheric pressure.
17	A preferred alkylation catalyst for this reaction is a
18	sulfonic acid catalyst, such as Amberlyst 15®, available
19	from Rohm and Haas, Philadelphia, Pennsylvania.
20	
21	Also contemplated for use in the present fuel additive
22	composition are the salts of the polyalkyl hydroxyaromatic
23	component, such as alkali metal, alkaline earth metal,
24	ammonium, substituted ammonium and sulfonium salts.
25	Preferred salts are the alkali metal salts of the polyalkyl
26	hydroxyaromatic compound, particularly the sodium and
27	potassium salts, and the substituted ammonium salts.
28	
29	
30	
31	
32	<u>Fuel Compositions</u>
33	
34	

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01 The fuel additive composition of the present invention will 02 generally be employed in a hydrocarbon distillate fuel 03 boiling in the gasoline or diesel range. The proper 04 concentration of this additive composition necessary in 05 order to achieve the desired detergency and dispersancy 06 varies depending upon the type of fuel employed, the 07 presence of other detergents, dispersants and other 80 additives, etc. Generally, however, from 150 to 7500 weight ppm, preferably from 300 to 2500 ppm, of the present 09 10 additive composition per part of base fuel is needed to 11 achieve the best results. 12 13 In terms of individual components, fuel compositions 14 containing the additive compositions of the invention will 15 generally contain about 50 to 2500 ppm of the aliphatic 16 amine and about 100 to 5000 ppm of the polyalkyl 17 hydroxyaromatic compound. The ratio of polyalkyl 18 hydroxyaromatic to aliphatic amine will generally range from about 0.5 to 10:1, and will preferably be about 2:1 or 19 20 greater. 21 22 The deposit control additive may be formulated as a 23 concentrate, using an inert stable oleophilic organic 24 solvent boiling in the range of about 150°F to 400°P. 25 Preferably, an aliphatic or an aromatic hydrocarbon solvent is used, such as benzene, toluene, xylene or higher-boiling 26 27 aromatics or aromatic thinners. Aliphatic alcohols of about 28 3 to 8 carbon atoms, such as isopropanol, isobutylcarbinol, 29 n-butanol and the like, in combination with hydrocarbon 30 solvents are also suitable for use with the detergent-31 dispersant additive. In the concentrate, the amount of the 32 present additive composition will be ordinarily at least 10% 33 by weight and generally not exceed 70% by weight, preferably 34

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1	10 to 50 weight percent and most preferably from 10 to 25
2	weight percent.
)3	
)4	In gasoline fuels, other fuel additives may also be included
)5	such as antiknock agents, e.g., methylcyclopentadienyl
)6	manganese tricarbonyl, tetramethyl or tetraethyl lead, or
7	other dispersants or detergents such as various substituted
8	amines, etc. Also included may be lead scavengers such as
9	aryl halides, e.g., dichlorobenzene or alkyl halides, e.g.,
LO	ethylene dibromide. Additionally, antioxidants, metal
L1	deactivators, pour point depressants, corrosion inhibitors
L2	and demulsifiers may be present.
13	
14	In diesel fuels, other well-known additives can be employed
15	such as pour point depressants, flow improvers, cetane
16	improvers, and the like.
17	
18	The following examples are presented to illustrate specific
19	embodiments of this invention and are not to be construed in
20	any way as limiting the scope of the invention.
21	
22	<u>examples</u>
23	
24	Example 1
25	
26	Preparation of Polyisobutyl Phenol
27	
28	To a flask equipped with a magnetic stirrer, reflux
29	condenser, thermometer, addition funnel and nitrogen inlet
30	was added 203.2 grams of phenol. The phenol was warmed to
31	40°C and the heat source was removed. Then,
32	73.5 milliliters of boron trifluoride etherate was added
33	dropwise. 1040 grams of Ultravis 10 polyisobutene
34	(molecular weight 950, 76% methylvinylidene, available from

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01	British Petroleum) was dissolv	ed in 1,863 millilit	ers of
02	hexane. The polyisobutene was	added to the reacti	on at a
03	rate to maintain the temperatu	re between 22-27°C.	The
04	reaction mixture was stirred f	or 16 hours at room	
05	temperature. Then, 400 millil	iters of concentrate	d ammonium
06	hydroxide was added followed b	y 2,000 milliliters	of hexane.
07	The reaction mixture was washe	d with water (3 x 2,	000
80	milliliters), dried over magne	sium sulfațe, filter	ed and the
09	solvents removed under vacuum	to yield 1,056.5 gra	uns of a
10	crude reaction product. The c	rude reaction produc	t was
11	determined to contain 80% of t	he desired product h	y proton
12	NMR and chromatography on sili	ca gel eluting with	hexane,
13	followed by hexane: ethylaceta	te: ethanol (93:5:2)	•
14			
15			
16	Exam	ple 2	
17			
18	Engin	<u>e Test</u>	
19			
20	A laboratory engine test was t		
21	valve and combustion chamber of		
22	additive composition of the in		_
23	4.3 liter, TBI (throttle body)
24	manufactured by General Motors	Corporation.	
25			
26	The major engine dimensions as	re listed below:	
27			_
28	Table I - Eng	ine Dimensions	
29			4
30	Bore	10.16 cm	
31	Stroke	8.84 cm	4
32	Displacement Volum		
33			41

34

9.3:1

The test procedure involves engine operation for 40 hours (24 hours a day) on a prescribed load and speed schedule representative of typical driving conditions. The cycle for engine operation during the test is as follows:

	Table II - Engine Driving Cycle					
Step	Time in Mode Dynamometer Speed [Sec]* Load [kg] [RPM]					
1	Idle	60	0	800		
2	City Cruise	150	10	1,500		
3	Acceleration	40	. 25	2,800		
4	Heavy HWY Cruise	210	15	2,200		
5	Light HWY Cruise	60	10	2,200		
6	Idle	60	0	800		
7	City Cruise	180	10	1,500		
8 ·	Idle	60	0	800		

All steps except step number 3, include a 15 second transition ramp. Step 3 include a 20 second transition ramp.

All of the test runs were made with the same base gasoline, which was representative of commercial unleaded fuel. The results are set forth in Table III.

31[°]

Table III Laboratory Engine Test Results				
Additive	Concentration,		Intake Valve Deposits, mg	Combustion Chamber Deposits, mg
Base Puel	•	Run 1 Run 2 Avg.	530 510 520	1,455 1,341 1,398
Amine/Neutral Oil*	200/800	Run 1 Run 2 Avg.	203 224 214	2,585 2,565 2,575
Polyalkyl Phenol ^b	400	Run 1 Run 2 Avg.	90 104 97	2,190 2,534 2,362
Amins/Polyalkyl Phenol*	200/400	Run 1 Run 2 Avg.	25 67 46	2,228 2,121 2,175

Mixture of 200 ppm polyisobutyl (MW=1300) ethylene diamine and 800 ppm of Chevron 500R neutral oil. The polyisobutyl group was derived from Parapol 1300 polyisobutene.

18 b Ultravis 10 polyisobutyl (MW = 950) phenol.

Mixture of 200 ppm polyisobutyl (MW = 1300) ethylene diamine and 400 ppm of Ultravis 10 polyisobutyl (MW = 950) phenol.

The results shown in Table III demonstrate that the combination of polyisobutyl phenol and polyisobutyl ethylene diamine has a synergistic effect and gives significantly better intake valve deposit control than either component by itself. Also, the addition of polyisobutyl phenol to the polyisobutyl ethylene diamine reduces the combustion chamber deposit weight compared to the polyisobutyl ethylene diamine alone.

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01	WHAT	IS C	LAIMEL) IS:
02		_		
03	1.	A fu	el add	litive composition comprising:
04			_	
05		(a)		el-soluble aliphatic amine selected from the
06			group	consisting of:
07				
08			(1)	A straight or branched chain hydrocarbyl-
09				substituted amine having at least one basic
10				nitrogen atom wherein the hydrocarbyl group
11				has a number average molecular weight of
12				about 250 to 3,000;
13				
14			(2)	A hydroxyalkyl-substituted amine comprising
15				the reaction product of (i) a polyolefin
16				epoxide derived from a branched-chain
17				polyolefin having a number average molecular
18				weight of about 250 to 3,000, and (ii) a
19				nitrogen-containing compound selected from
20				ammonia, a monoamine having from 1 to 40
21		-		carbon atoms, and a polyamine having from 2
22				to about 12 amine nitrogen atoms and from 2
23		•		to about 40 carbon atoms; and
24				
25			(3)	A straight or branched chain hydrocarbyl-
26				substituted succinimide comprising the
27				reaction product of a straight or branched
28				chain hydrocarbyl-substituted succinic acid
29				or anhydride, wherein the hydrocarbyl group
30				has a number average molecular weight of
31				about 250 to 3,000, and a polyamine having
32				from 2 to about 12 amine nitrogen atoms and 2
33				to about 40 carbon atoms; and

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		·
01		(b) A polyalkyl hydroxyaromatic compound or salt
02		thereof wherein the polyalkyl group has sufficient
03		molecular weight and carbon chain length to render
04		the polyalkyl hydroxyaromatic compound soluble in
05 ·		hydrocarbons boiling in the gasoline or diesel
06	•	range.
07		
08	2.	The fuel additive composition according the Claim 1,
09		wherein the hydrocarbyl or hydroxyalkyl substituent on
10		the aliphatic amine of component (a) has a number
11		average molecular weight of about 700 to 2,200.
12		
13	3.	The fuel additive composition according to Claim 2,
14		wherein the hydrocarbyl or hydroxyalkyl substituent on
15		the aliphatic amine of component (a) has a number
16		average molecular weight of about 900 to 1,500.
17		•
18	4.	The fuel additive composition according to Claim 1,
19		wherein the aliphatic amine of component (a) is a
20		straight or branched chain hydrocarbyl-substituted
21		amine.
22		
23	5.	The fuel additive composition accord to Claim 4,
24		wherein the aliphatic amine of component (a) is a
25		branched chain hydrocarbyl-substituted amine.
26		
27	6.	The fuel additive composition according to Claim 5,
28	•	wherein the aliphatic amine of component (a) is a
29		polyisobutyl amine.
30		rant enamed a museum
31	7.	The fuel additive composition according to Claim 4,
32	, ,	wherein the amine moiety of the aliphatic amine is
33		derived from a polyamine having from 2 to 12 amine
33		derived from a bolyamine having from 2 to 12 amine

nitrogen atoms and from 2 to 40 carbon atoms.

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01	8.	The fuel additive composition according to claim 7,
02		wherein the polyamine is a polyalkylene polyamine
03		having 2 to 12 amine nitrogen atoms and 2 to 24 carbon
04		atoms.
05		
06	9.	The fuel additive composition according to Claim 8,
07		wherein the polyalkylene polyamine is selected from the
80		group consisting of ethylene diamine, diethylene
09		triamine, triethylene tetramine and tetraethylene
10		pentamine.
11		
12	10.	The fuel additive composition according to Claim 9,
13		wherein the polyalkylene polyamine is ethylene diamine
14		or diethylene triamine.
15		
16	11.	The fuel additive composition according to Claim 1,
17		wherein the aliphatic amine of component (a) is a
18		hydroxyalkyl-substituted amine.
19		
20.	12.	The fuel additive composition according to Claim 11,
21		wherein the hydroxyalkyl-substituted amine is derived
22		from a branched chain polyolefin selected from
23		polypropylene or polyisobutene.
24		
25	13.	The fuel additive composition according to Claim 12,
2 6		wherein the branched chain polyolefin is polyisobutene.
27		
28	14.	The fuel additive composition according to Claim
29		11, wherein the hydroxyalkyl-substituted amine is
30		derived from a polyamine having from 2 to about 12
31		amine nitrogen atoms and 2 to about 40 carbon atoms.
32		

15. The fuel additive composition according to Claim 14,

wherein the polyamine is a polyalkylene polyamine

33

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01 02 03 04		wherein the alkylene group contains from 2 to 6 carbon atoms and the polyalkylene polyamine contains from 2 to 12 nitrogen atoms and from 2 to 24 carbon atoms.
05 06 07 08 09	16.	The fuel additive composition according to Claim 15, wherein the polyalkylene polyamine is selected from the group consisting of ethylene diamine, polyethylene polyamine, propylene diamine and polypropylene polyamine.
11 12 13 14	17.	The fuel additive composition according to Claim 1, wherein the aliphatic amine of component (a) is a straight or branched chain hydrocarbyl-substituted succinimide.
16 17 18	18.	The fuel additive composition according to Claim 17, wherein the aliphatic amine is a branched chain hydrocarbyl-substituted succinimide.
20 21 22 23	19.	The fuel additive composition according to Claim 18, wherein the branched chain hydrocarbyl substituent is polyisobutyl.
24 25 26 27 28	20.	The fuel additive composition according to Claim 17, wherein the hydrocarbyl-substituted succinimide is derived from a polyalkylene polyamine having 2 to 12 amine nitrogen atoms and 2 to 24 carbon atoms.
29 30 31 32	21.	The fuel additive composition according to Claim 20, wherein the polyalkylene polyamine is selected from the group consisting of ethylene diamine, diethylene triamine, triethylene tetramine and tetraethylene pentamine.

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01 02 03 04	22.	The fuel additive composition according to Claim 21, wherein the polyalkylene polyamine is ethylene diamine or diethylene triamine.
05 06 07 08 09	23.	The fuel additive composition according to Claim 1, wherein the polyalkyl hydroxyaromatic compound of component (b) has a polyalkyl group with an average molecular weight of about 400 to 5,000.
10 11 12	24.	wherein the hydroxyaromatic compound is phenol.
13 14 15 16 17	25.	The fuel additive composition according to Claim 1, wherein the polyalkyl substituent in component (b) is derived from polypropylene, polybutylene, or polyalphaolefin oligomers of 1-decene.
18 19 20 21	26.	The fuel additive composition according to Claim 25, wherein the polyalkyl substituent in component (b) is derived from polyisobutylene.
22 23 24 25	27.	The fuel additive composition according to Claim 26, wherein the polyisobutylene contains at least about 20% of a methylvinylidene isomer.
26 27 28 29 30	28.	The fuel additive composition according to Claim 1, wherein component (a) is a polyisobutyl amine, wherein the amine moiety is derived from ethylene diamine or diethylene triamine, and component (b) is a polyisobutyl phenol.
32 33 34	29.	A fuel composition comprising a major amount of hydrocarbons boiling in the gasoline or diesel range

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01	and an e	effective detergent amount of an additive
02	composit	ion comprising:
03		
04	(a) Af	uel-soluble aliphatic amine selected from the
05		oup consisting of
06	•	
07	(1)	A straight or branched chain hydrocarbyl-
08		substituted amine having at least one basic
09		nitrogen atom wherein the hydrocarbyl group
10		has a number average molecular weight of
11		about 250 to 3,000;
12		·
13	(2)	A hydroxyalkyl-substituted amine comprising
14		the reaction product of (i) a polyolefin
15		epoxide derived from a branched-chain
16		polyolefin having a number average molecular
17		weight of about 250 to 3,000, and (ii) a
18		nitrogen-containing compound selected from
19		ammonia, a monoamine having from 1 to 40
20		carbon atoms, and a polyamine having from 2
21		to about 12 amine nitrogen atoms and from 2
22		to about 40 carbon atoms; and
23		
24	(3)	A straight or branched chain hydrocarbyl-
25		substituted succinimide comprising the
26		reaction product of a straight or branched
27		chain hydrocarbyl-substituted succinic acid
8		or anhydride, wherein the hydrocarbyl group
29		has a number average molecular weight of
30		about 250 to 3,000, and a polyamine having
11		from 2 to about 12 amine nitrogen atoms and 2
12		to about 40 carbon atoms; and
3		

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01		(b)	a po	lyalkyl hydroxyaromatic compound or salt
02			ther	eof wherein the polyalkyl group has sufficient
03			mole	cular weight and carbon chain length to render
04			the	polyalkyl hydroxyaromatic compound soluble in
05	•		hydr	ocarbons boiling in the gasoline or diesel
06			rang	e.
07				
80	30.	A fu	el co	ncentrate comprising an inert stable
09		oleo	phili	c organic solvent boiling in the range of from
10		abou	t 150	°F to 400°F and from about 10 to 70 weight
11		perc	ent o	f an additive composition comprising:
12				
13		(a)	A fu	el-soluble aliphatic amine selected from the
14			grou	p consisting of
15				
16			(1)	A straight or branched chain hydrocarbyl-
17				substituted amine having at least one basic
18				nitrogen atom wherein the hydrocarbyl group
19				has a number average molecular weight of
20				about 250 to 3,000;
21				
22			(2)	A hydroxyalkyl-substituted amine comprising
23				the reaction product of (i) a polyolefin
24				epoxide derived from a branched-chain
25				polyolefin having a number average molecular
26				weight of about 250 to 3,000, and (ii) a
27				nitrogen-containing compound selected from
28				ammonia, a monoamine having from 1 to 40
29				carbon atoms, and a polyamine having from 2
30				to about 12 amine nitrogen atoms and from 2
31				to about 40 carbon atoms; and
32		•		•
33			(3)	A straight or branched chain hydrocarbyl-
3 4				substituted succinimide comprising the

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01		reaction product of a straight or branched.
02		chain hydrocarbyl-substituted succinic acid
03		or anhydride, wherein the hydrocarbyl group
04		has a number average molecular weight of
05		about 250 to 3,000, and a polyamine having
06		from 2 to about 12 amine nitrogen atoms and 2
07		to about 40 carbon atoms; and
08		, and
09	(b)	a polyalkyl hydroxyaromatic compound or salt
10	(,	thereof wherein the polyalkyl group has sufficient
11		molecular weight and carbon chain length to render
12		the polyalkyl hydroxyaromatic compound soluble in
13		hydrocarbons boiling in the gasoline or diesel
14		range.
15		lange.
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INTERNATIONAL SEARCH REPORT

International application No. PCT/US93/12555

	SSIFICATION OF SUBJECT MATTER	•		
US CL	IPC(5) :C10L 1/18, 1/22, 10/00 US CL :44/347, 412, 432, 433, 434, 450			
	to International Patent Classification (IPC) or to both	national classification and IPC		
	LDS SEARCHED secumentation searched (classification system follows)	ed bu placeification combale)		
	•	co oy classification symbols;		
U.S. :	44/347, 412, 432, 433, 434, 450			
Documenta	tion searched other than minimum documentation to t	he extent that such documents are included	in the fields searched	
Electronic o	data base consulted during the international scarch (r	name of data base and, where practicable	, search terms used)	
C. DOC	CUMENTS CONSIDERED TO BE RELEVANT		·	
Category*	Citation of document, with indication, where a	appropriate, of the relevant passages	Relevant to claim No.	
Y	US, A, 3,849,085 (KREUZ ET A (SEE ENTIRE DOCUMENT)	L.) 19 NOVEMBER 1974,	1-30	
Y	US, A, 4,123,232 (FROST, J (COLUMN 1, LINES 30-37, COLU		1-16, 23-30	
P,Y	US, A, 5,192,335 (CHERPECK) ABSTRACT)	09 MARCH 1993, (SEE	1-30	
Y	US, A, 5,114,435 (ABRAMO (COLUMN 2, LINES 33-44) C (COLUMN 3, LINE 62)	ET AL.) 19 MAY 1992 OLUMN 2, LINE 48 TO	1-10, 17-30	
Y	UK, A, 2,156,848 A (EX ENGINEERING COMPANY) 16 OC DOCUMENT)		1-30	
X Furth	ner documents are listed in the continuation of Box (C. See patent family annex.		
	Special extraories of cited documents: T later document published after the international filing date or priority date and not in conflict with the application but cited to understand the			
"A" do	cument defining the general state of the art which is not considered be of particular relevance	principle or theory underlying the inve		
'E' ear	tier document published on or after the international filing date	"X" document of particular relevance; the considered novel or cannot be consider	red to involve en inventive exp	
citz	cument which may throw doubts on priority chaim(s) or which is of to establish the publication data of another citation or other cital reason (as meeting)	"Y" document of particular relevance; the	chimal invention cannot be	
.O. qoo	cument referring to an oral disclosure, use, exhibition or other	considered to involve an inventive combined with one or more other such	step when the document in documents, such combination	
P doc	nes rument published prior to the international filling date but later than priority data channel	being obvious to a person skilled in the '&' document member of the same patent		
	actual completion of the international search	Date of mailing of the international sea	rch report	
17 March	1994	30 MAR 1994	1	
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INTERNATIONAL SEARCH REPORT

International application No. PCT/US93/12555

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	ion). DOCUMENTS CONSIDERED TO BE RELEVANT	
Category	Citation of document, with indication, where appropriate, of the relevant passage	Relevant to claim No
A	US, A, 3,438,757 (HONNEN ET AL.) 15 APRIL 1969	
A.	US, A, 4,832,702 (KUMMER ET AL.) 23 MAY 1989	
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